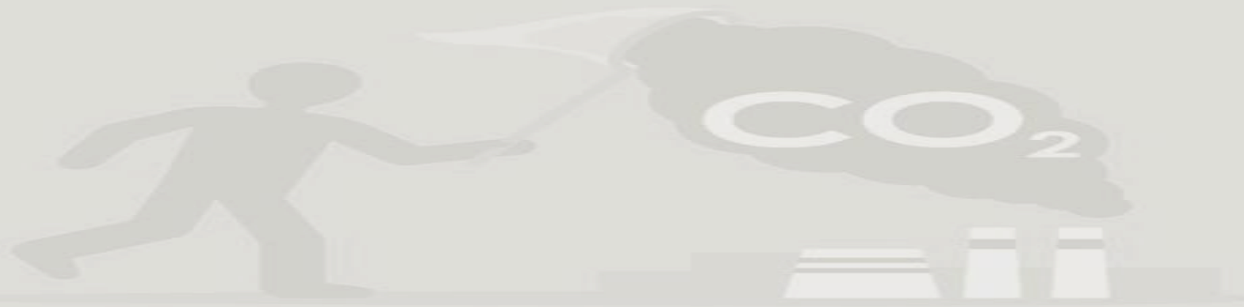




Enhanced ex-situ biomethanation of H₂ and CO₂ for biomethane production using graphene oxide coated carrier



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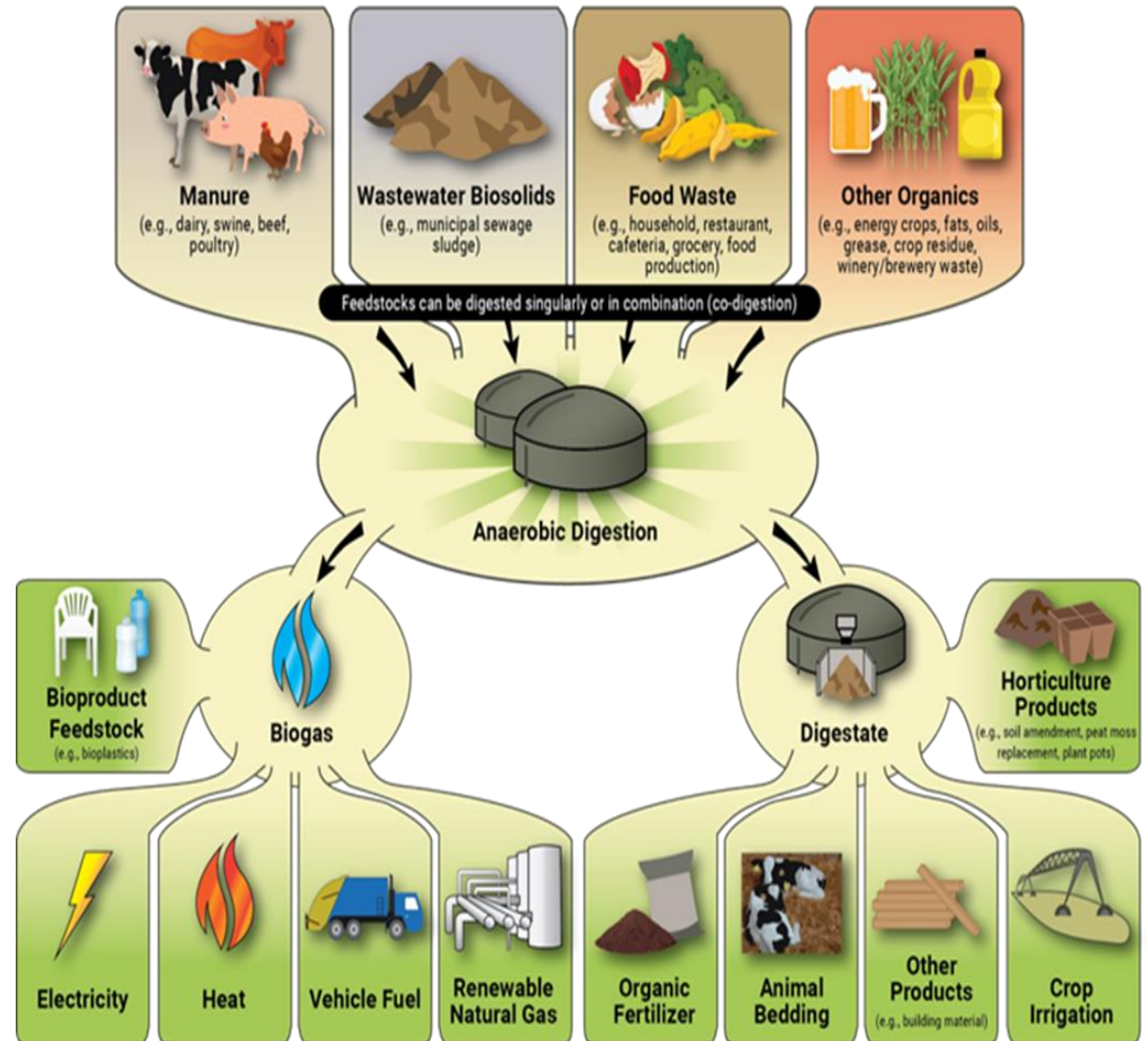
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Background

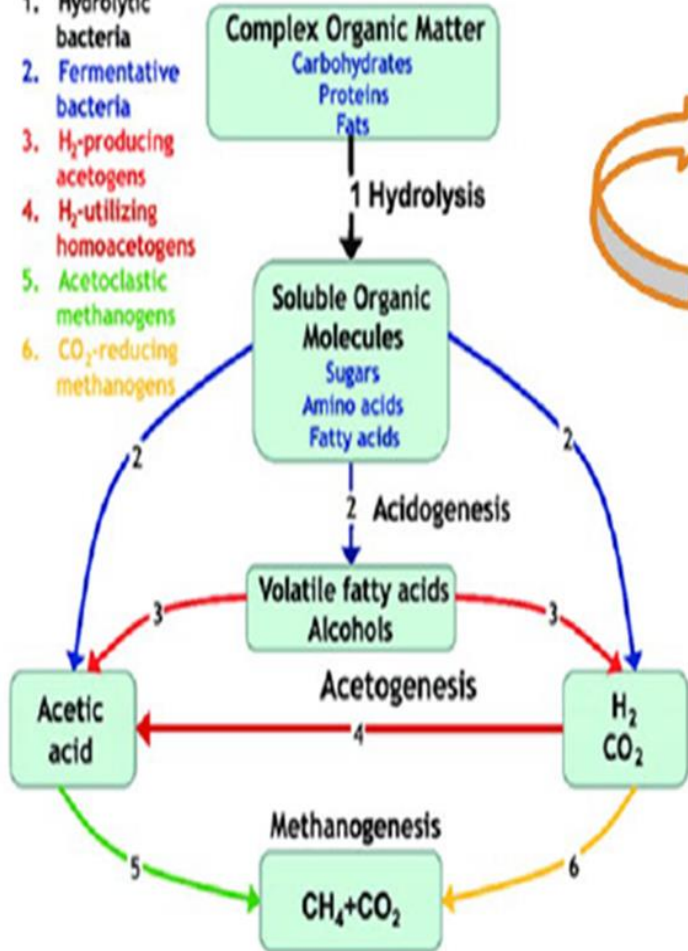
Biogas- a mixture of mainly methane (CH_4) and carbon dioxide (CO_2) produced during anaerobic digestion of organic wastes.



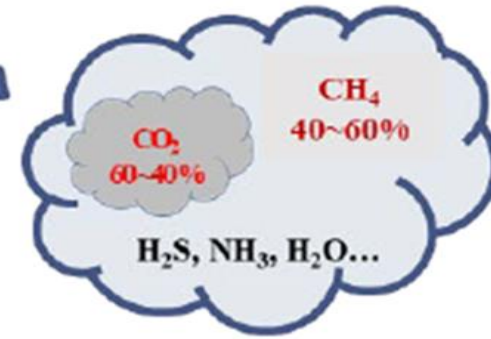
Background

Anaerobic digestion

1. Hydrolytic bacteria
2. Fermentative bacteria
3. H₂-producing acetogens
4. H₂-utilizing homoacetogens
5. Acetoclastic methanogens
6. CO₂-reducing methanogens



Biogas plant



- ✓ Low CH₄ content
- ✓ Low calorific value
- ✓ Limit use to heat and electricity

❖ What is Biogas Upgrading?

Removal or conversion of CO₂ contained in raw biogas to increase the methane content (> 95%) of the final output gas

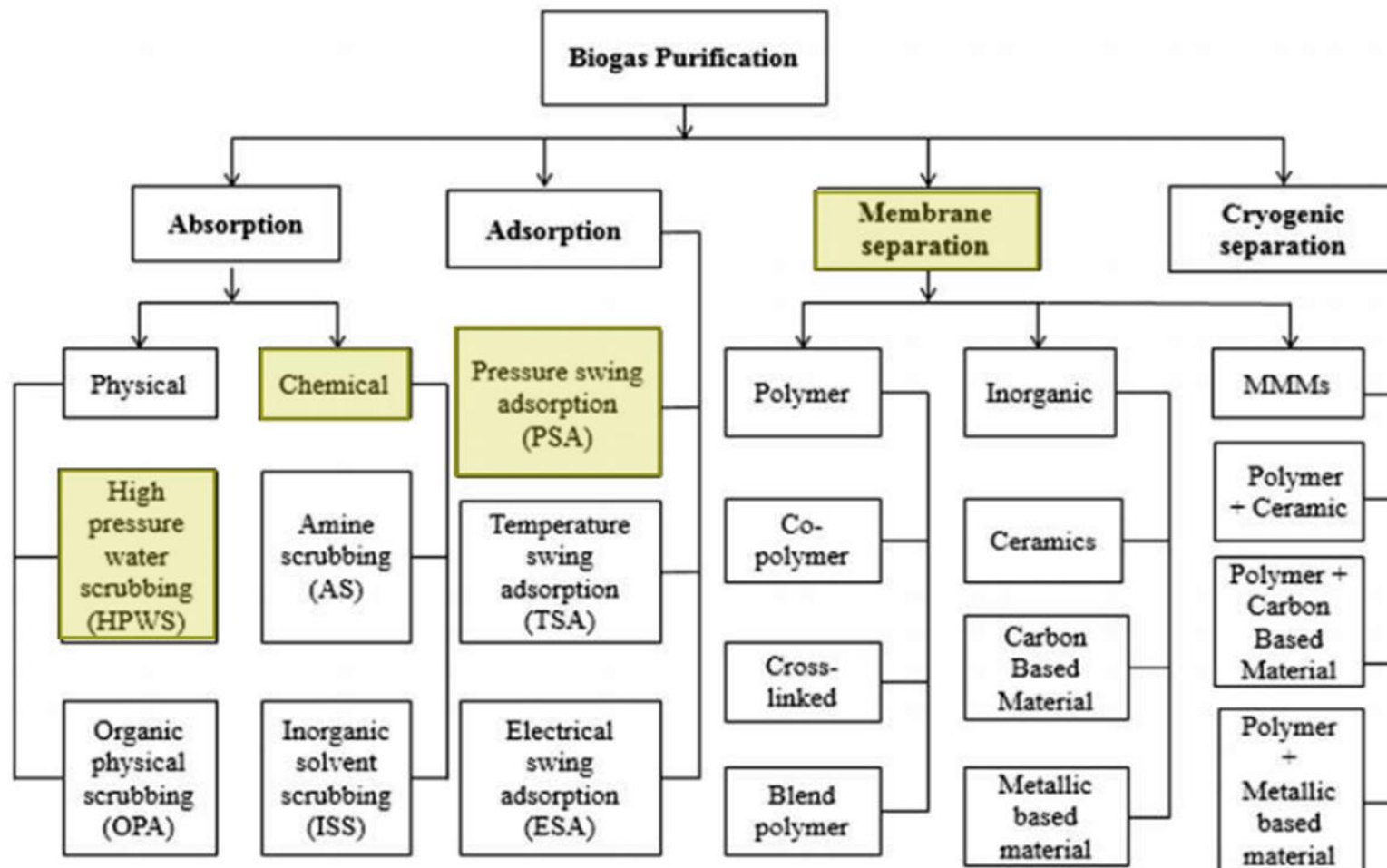


Why biogas upgrading?

- ❖ Higher CH₄ content
- ❖ Decrease CO₂ emission
- ❖ Inject into natural gas grid
- ❖ Utilize as transport fuel

Background

Physicochemical biogas upgrading technologies



Background

- Currently biogas upgrading technologies are expensive and energy intensive (pressure, chemicals or membrane)
- Upgrading for small and medium scale facilities is not economically feasible
- Losses of CO₂ and CH₄

Efficiency of main physicochemical biogas upgrading technologies

	Cryogenic	Sabatier Process	PSA	Water Scrubbing	Physical Scrubbing	Chemical Absorption	Membrane Separation
Consumption for raw biogas (kWh/Nm ³)	0.76	nf	0.23-0.30	0.25-0.3	0.2-0.3	0.05-0.15	0.18-0.20
Consumption for clean biogas (kWh/Nm ³)	nf	nf	0.29-1.00	0.3-0.9	0.4	0.05-0.25	0.14-0.26
Heat consumption (kWh/Nm ³)	nf	nf	None	None	<0.2	0.5-0.75	None
Heat demand (°C)	-196	270			55-80	100-180	
Cost	High	Medium	Medium	Medium	Medium	High	High
CH ₄ losses (%)	2	nf	<4	<2	2-4	<0.1	<0.6
CH ₄ recovery (%)	97-98	97-99	96-98	96-98	96-98	96-99	96-98
Prepurification	Yes	Recommended	Yes	Recommended	Recommended	Yes	Recommended
H ₂ S co-removal	Yes	No	Possible	Yes	Possible	Contaminant	Possible
N ₂ and O ₂ co-removal	Yes	No	Possible	No	No	No	Partial
Operation pressure (bar)	80	8-10	3-10	4-10	4-8	Atmospheric	5-8
Pressure at outlet (bar)	8-10		4-5	7-10	1.3-7.5	4-5	4-6

Background

- ❖ Fluctuating production of electricity from solar panels or wind turbines
- ❖ Excess electricity requires solutions
- ❖ Difficult to store electricity

The solution



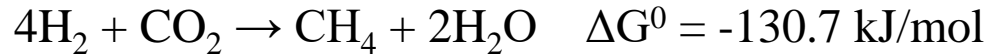
- ❖ Excess electricity can be converted to biomethane by ex-situ bimethanation
- ❖ Easy to store and transport biomethane where access to natural gas grid

Ex-situ biomethanation

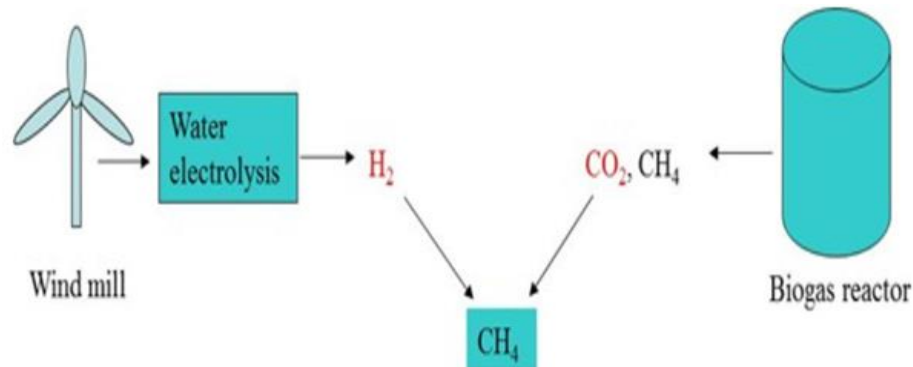
Developed by **Gang Luo**, Postdoc, and **Irini Angelidaki**, Professor

The Principle:

CO₂ together with H₂ could be used by hydrogenotrophic methanogens for CH₄ production



- H₂ as electron donor and CO₂ as electron acceptor and carbon source
- H₂ could be obtained by water electrolysis using surplus renewable electricity (e.g., wind mill, solar panel)
- Power-to-Gas (**PtG**)

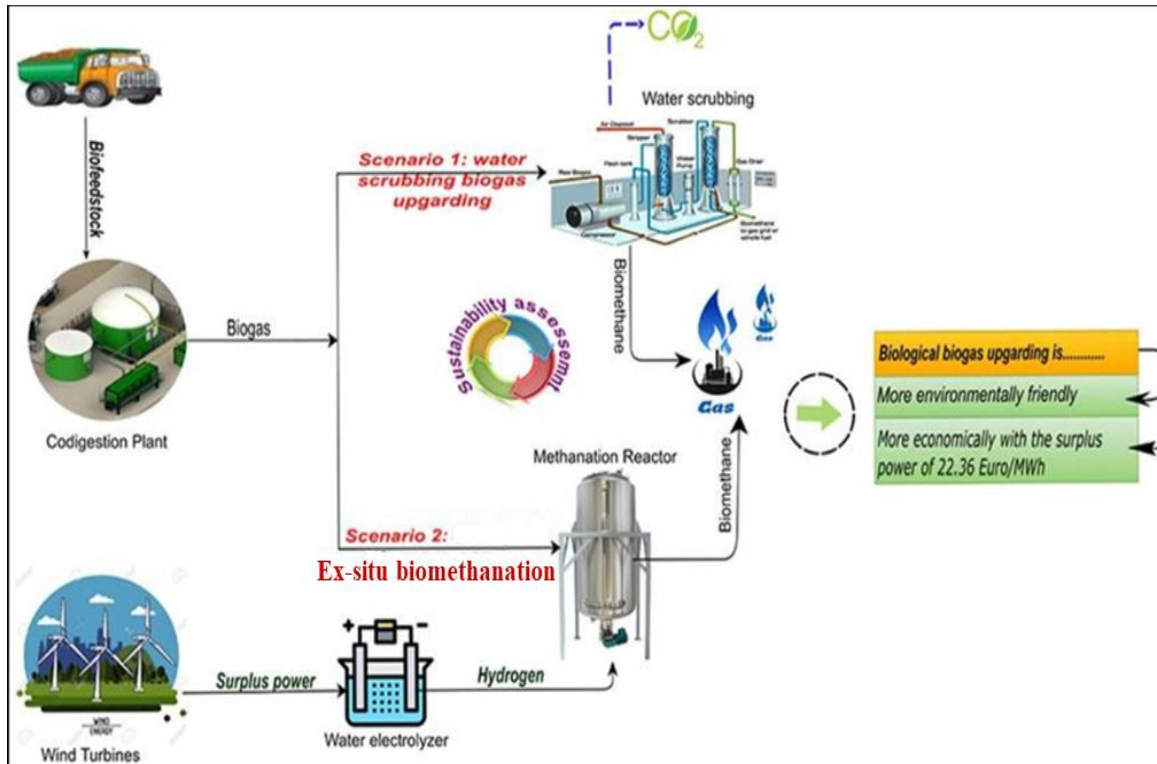


Ex-situ biomethanation

Why ex-situ biomethanation

?

- ❖ Operation at ambient conditions (atmospheric pressure and moderate temperature) without use of catalyst and chemicals
- ❖ Utilization of CO₂ instead of removal to increase CH₄ content in output gas
- ❖ Higher resistance to gas impurities like H₂S, organic acids, NH₄

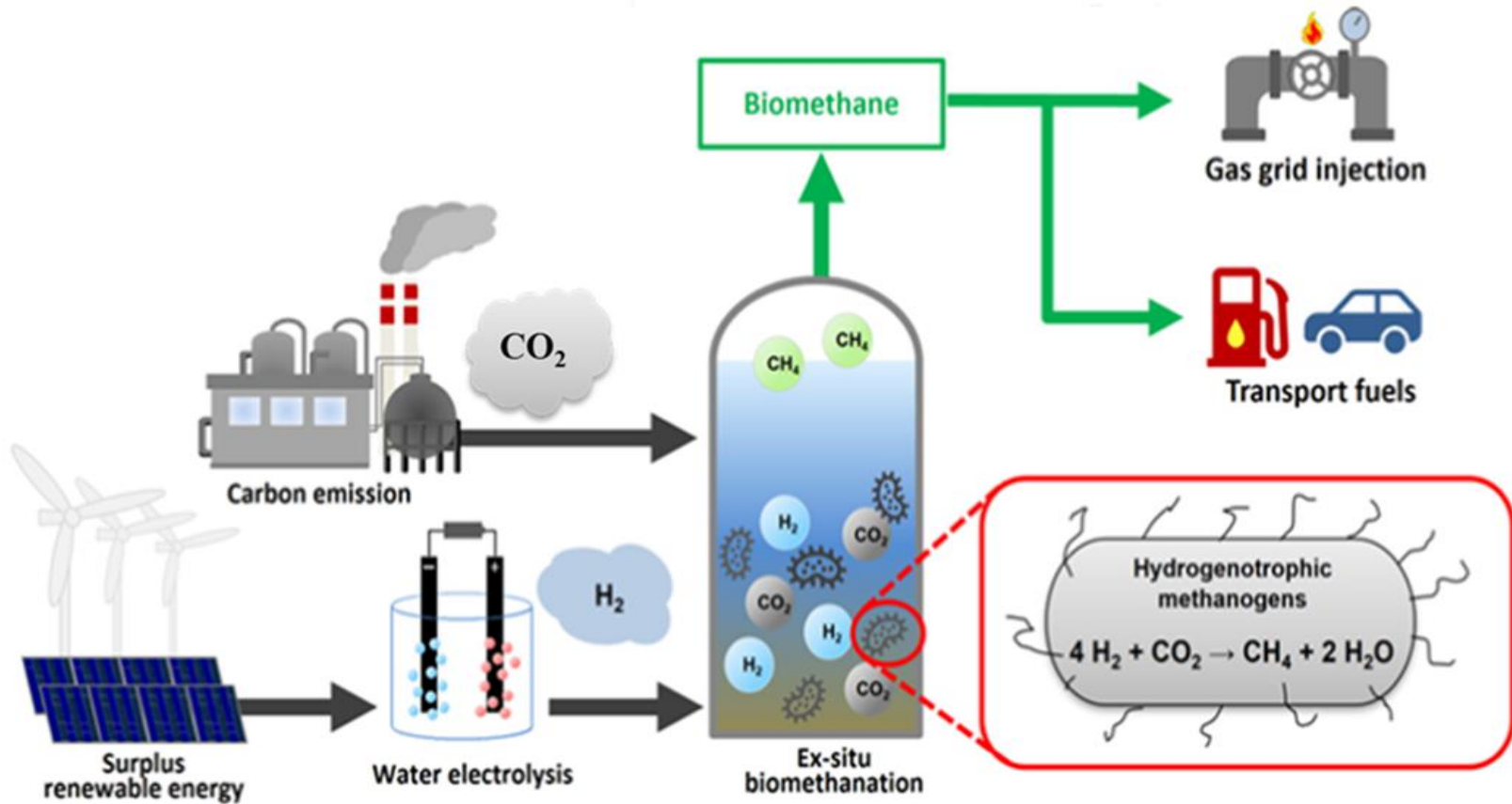


Why biomethanation?

- Eco-friendly
- Cost-effective
- Low energy demanding
- Utilization of valuable CO₂

Background

Ex-situ biomethanation with Power-to-gas concept for biomethane production



❖ **H₂ production** using surplus renewable electricity (windmills or solar panels) through **water electrolysis**

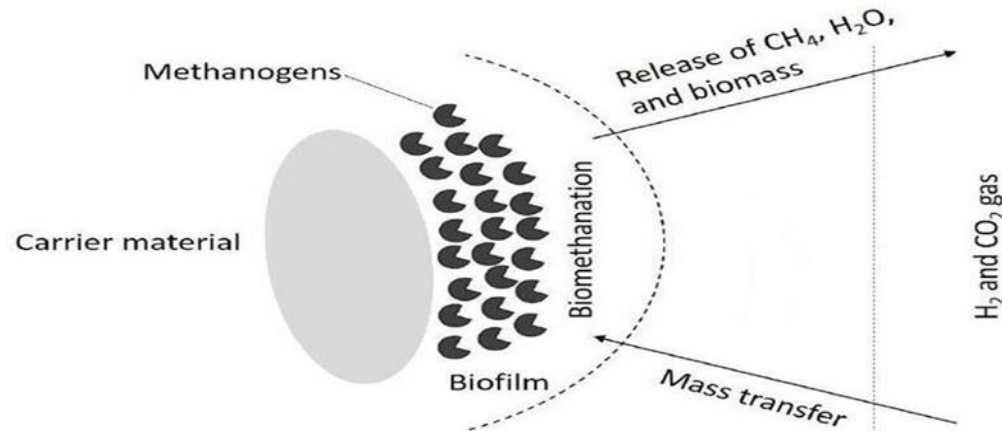
❖ Power-to-Gas (P2G)

Research gaps

- ❖ Low H_2 gas-liquid mass transfer
- ❖ Biofilm detachment under high gas loading rate (H_2/CO_2)
- ❖ Dilution of methanogens (from metabolic H_2O production during hydrogenotrophic methanogenesis)
- ❖ Stability of ex-situ biomethanation process under intermittent gas supply

Hypothesis

- ◆ Carrier material coated by conductive material (e.g., graphene oxide) can offer various benefits:
- ❖ Robust and dense biofilm formation (High microbial biomass)
- ❖ High specific surface area for microbes (biofilm)-gas-liquid phase interaction, which ultimately increase H_2 gas-liquid mass transfer and enhance biomethane production
- ❖ Prevent risk of dilution of methanogen from metabolic water production (No loss of microbes)
- ❖ Prevent loss of conductive materials



Objective

Main objective

- To enhance biomethane production using ex-situ biomethanation trickle bed reactor setup

Specific objective

- To investigate performance of graphene oxide coated carrier materials for H₂ gas-liquid mass transfer and biomethane production & apply in trickle bed reactor setup

Experimental setup

Preparation of coated carrier materials for biomethane production

- ❖ Graphene oxide (GO) was prepared by Hummer's method with some modifications.
- ❖ GO coated carrier was prepared from the solution deposition method.



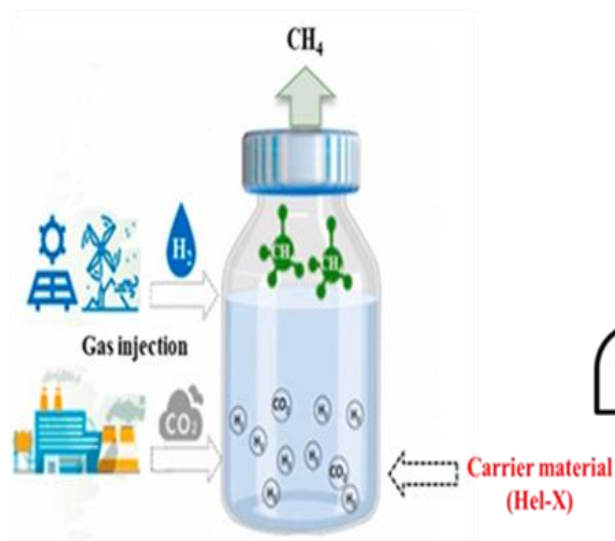
Hel-X carriers



GO-coated Hel-X carriers

Fig. Photos of conventional Hel-X carriers and graphene oxide (GO)-coated Hel-X carriers

Experimental setup



Serum bottle



Serum bottles kept at shaking incubator
(37° C, 150 rpm)



Analytical equipment



Gas chromatography



HPLC



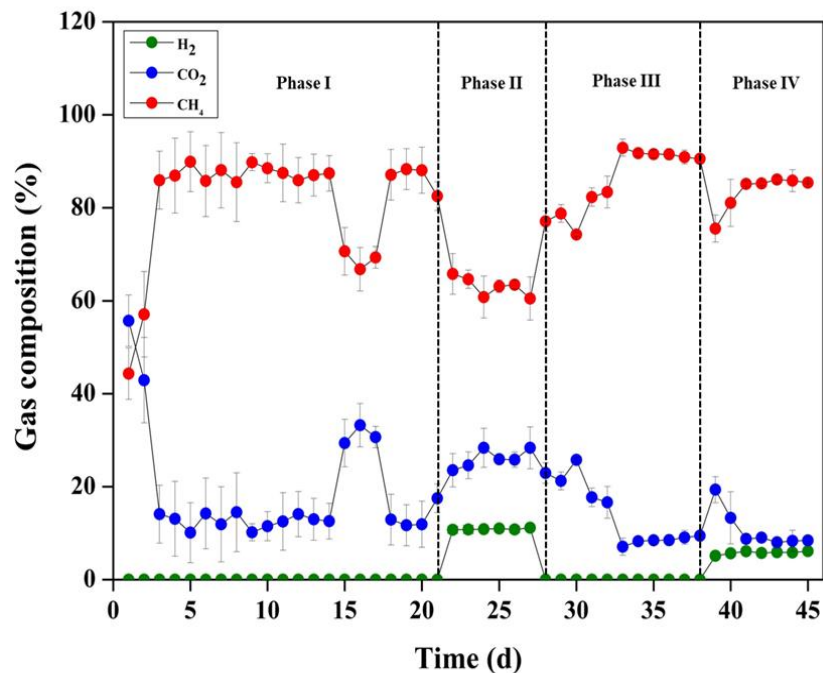
Hel-X carriers



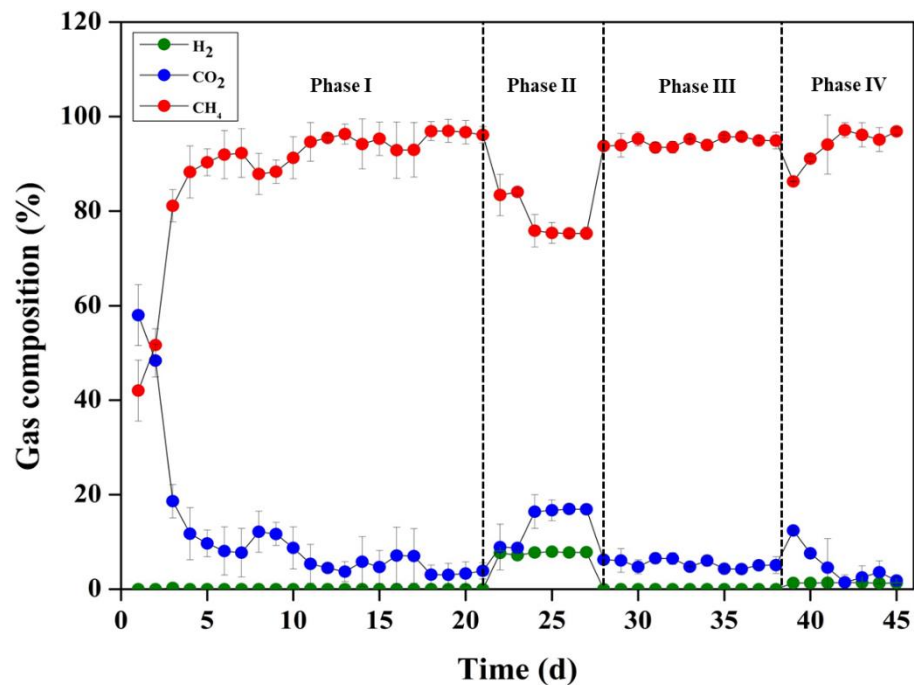
GO-coated Hel-X carriers

Experimental results

a) Control reactor (CR)



b) GO-coated reactor



- ❖ CH₄ content reached >98% in GO coated reactor
- ❖ Dramatic reduction in CH₄ content in CR than GO under stressful condition (i.e., high H₂ loading rate)
- ❖ Complete conversion of H₂ for reducing CO₂ to CH₄

Figure. Biogas composition of **a)** Control reactor filled with Hel-X carrier and **b)** Reactor filled with graphene oxide (GO)-coated Hel-X carrier. The lines represent mean values (n=3) and error bars denote the standard deviation.

Experimental results

Table. Reactors performance at different phases under steady-state conditions

Phase	I		II		III		IV	
Parameter	CR	GO	CR	GO	CR	GO	CR	GO
H ₂ loading rate (L/L _R .d)		3.2		4.8		3.2		3.2
CO ₂ loading rate (L/L _R .d)		0.8		1.2		1.2		1.2
Gas retention time (h)		24		24		24		18
CH₄ production rate (L/L_R.d)	0.46 ± 0.08	0.65 ± 0.07	0.56 ± 0.07	0.68 ± 0.11	0.53 ± 0.03	0.69 ± 0.01	0.50 ± 0.02	0.69 ± 0.02
Biogas composition (%)								
H ₂	0.00 ± 0.00	0.00 ± 0.00	10.98 ± 0.19	7.83 ± 0.19	0.00 ± 0.00	0.00 ± 0.00	5.99 ± 0.28	1.31 ± 0.08
CO ₂	13.31 ± 5.51	4.93 ± 2.81	26.67 ± 2.40	16.84 ± 1.58	9.27 ± 1.25	5.08 ± 1.10	8.39 ± 1.49	2.69 ± 1.52
CH₄	86.69 ± 5.52	95.07 ± 2.84	62.35 ± 2.31	75.33 ± 1.77	90.73 ± 1.27	94.92 ± 1.12	85.62 ± 1.27	96.00 ± 1.60
ηH ₂ (%)	100 ± 0.00	100 ± 0.00	97.96 ± 0.07	98.52 ± 0.06	100 ± 0.00	100 ± 0.00	98.91 ± 0.04	99.71 ± 0.05

- ❖ **GO-coated reactor** performed better in terms of **CH₄ production rate (38% ↑) & CH₄ content (13% ↑)** than **control reactor (CR)**
- ❖ **GO-coated reactor** showed **resilience** even under 18 h gas retention time than CR
- ❖ GO facilitated **Direct interspecies electron transfer (DIET) for high CH₄ content due to its high electrical conductivity and large specific surface area**

Conclusion

- ❖ Performance of GO-coated Hel-X carriers were evaluated for ex-situ biomethanation
- ❖ **GO-coated reactor** performed better in terms of **CH₄ production rate & CH₄ content** than **control reactor**
- ❖ GO-coated carrier enhanced gas conversion efficiency and increased CH₄ production
- ❖ GO act as an **electron mediator**, accelerated **DIET** for enhanced bioconversion of CO₂ and H₂
- ❖ Coupling of H₂ and CO₂ in **GO-coated reactor** met **natural gas quality CH₄ content (>95%)**

THANK YOU
MERCI